

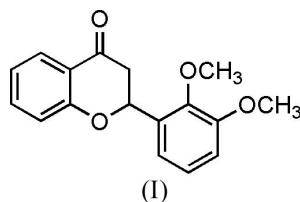
2-(2,3-Dimethoxyphenyl)chroman-4-one

Hui Wu,^{a*} Zhou Xu,^a Jun Zhou^a
and Yong-Min Liang^b^aDepartment of Chemistry, Xuzhou Normal University, Xuzhou, Jiangsu, 221116, People's Republic of China, and ^bDepartment of Chemistry, Lanzhou University, State Key Laboratory of Applied Organic Chemistry, Lanzhou, Gansu, 730000, People's Republic of ChinaCorrespondence e-mail:
wuhui72@yahoo.com.cn

Key indicators

Single-crystal X-ray study
 $T = 298$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.051
 wR factor = 0.126
Data-to-parameter ratio = 12.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.The title compound, $\text{C}_{17}\text{H}_{16}\text{O}_4$, was synthesized by the reaction of 2,3-dimethoxybenzaldehyde and 2-hydroxyacetophenone in 20% NaOH solution for 2 h, followed by crystallization from ethyl acetate. The pyrone ring adopts a chair conformation.

Comment

Flavanones are present in a wide variety of natural products exhibiting interesting biological activities (Heinisch & Holzer, 1991; Salvatore *et al.*, 1998). Thus, the synthesis of flavanones and their derivatives is of great interest in organic chemistry. It is found from biological examination that the flavanone (I) has good molluscicidal activity, and we report its crystal structure here. In (I), the pyrone ring is slightly distorted and adopts a chair conformation (Fig. 1). The bond lengths and angles in (I) show normal values (Table 1).

Experimental

The title compound, (I), was prepared from 2,3-dimethoxybenzaldehyde (2 mmol, 0.033 g) and 2-hydroxyacetophenone (2 mmol, 0.027 g) in 20% NaOH solution by stirring at room temperature for 2 h (yield 90%, m.p. 365–366 K). Crystals of (I) suitable for X-ray diffraction were obtained by crystallization from ethyl acetate.

Crystal data

$\text{C}_{17}\text{H}_{16}\text{O}_4$	$Z = 2$
$M_r = 284.30$	$D_x = 1.344$ Mg m ⁻³
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 7.547$ (3) Å	Cell parameters from 732 reflections
$b = 9.122$ (2) Å	$\theta = 2.5$ – 22.9°
$c = 11.437$ (3) Å	$\mu = 0.10$ mm ⁻¹
$\alpha = 107.905$ (2) $^\circ$	$T = 298$ (2) K
$\beta = 108.506$ (4) $^\circ$	Block, colorless
$\gamma = 92.139$ (3) $^\circ$	$0.48 \times 0.34 \times 0.18$ mm
$V = 702.4$ (4) Å ³	

Data collection

Bruker SMART CCD area-detector diffractometer	2443 independent reflections
φ and ω scans	1262 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.027$
$T_{\text{min}} = 0.956$, $T_{\text{max}} = 0.983$	$\theta_{\text{max}} = 25.0^\circ$
3693 measured reflections	$h = -8 \rightarrow 8$
	$k = -10 \rightarrow 9$
	$l = -13 \rightarrow 13$

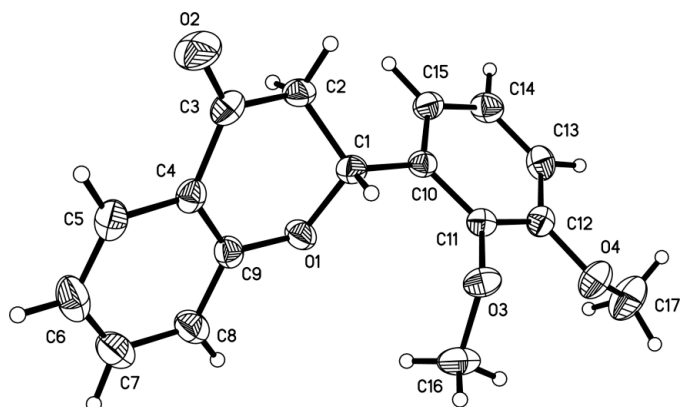


Figure 1
The molecular structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme.

Refinement

Refinement on F^2

$$R[F^2 > 2\sigma(F^2)] = 0.051$$

$$wR(F^2) = 0.126$$

$$S = 1.00$$

2443 reflections

192 parameters

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.051P)^2]$$

$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\max} < 0.001$$

$$\Delta\rho_{\max} = 0.19 \text{ e } \text{Å}^{-3}$$

$$\Delta\rho_{\min} = -0.18 \text{ e } \text{Å}^{-3}$$

Table 1

Selected geometric parameters (Å, °).

O1—C1	1.458 (3)	C1—C10	1.500 (3)
O2—C3	1.207 (3)	C1—C2	1.502 (3)
C9—O1—C1	114.26 (19)	O1—C1—C2	109.0 (2)
O1—C1—C10	105.98 (18)	C10—C1—C2	117.5 (2)
C9—O1—C1—C10	178.8 (2)	C1—C2—C3—O2	143.9 (3)
O1—C1—C2—C3	63.2 (3)	O1—C1—C10—C11	-75.1 (3)
C10—C1—C2—C3	-176.3 (2)	O1—C1—C10—C15	103.3 (3)

All H atoms were placed in calculated positions and treated using a riding model, with C—H distances of 0.93–0.98 Å and $U_{\text{iso}}(\text{H})$ values of $1.2U_{\text{eq}}(\text{C})$ [$1.5U_{\text{eq}}(\text{C})$ for methyl groups].

Data collection: *SMART* (Siemens, 1996); cell refinement: *SMART*; data reduction: *SAINT* (Siemens, 1996); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics:

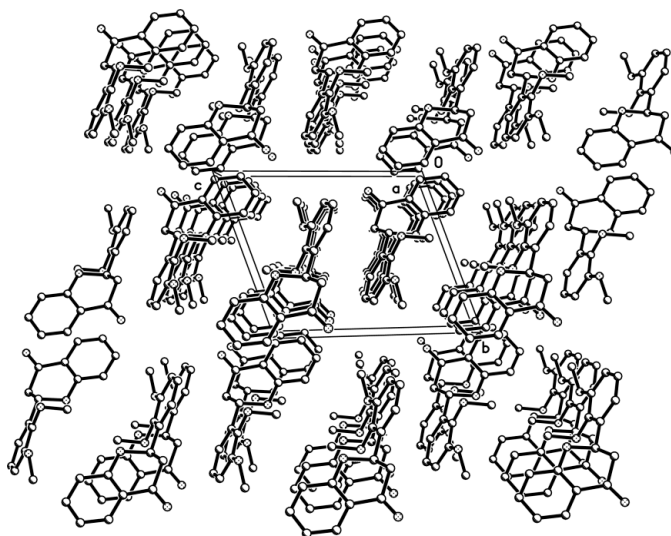


Figure 2

The molecular packing of (I). H atoms have been omitted.

SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: *SHELXTL*.

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